

Compressive deformation behavior of ternary compound Cr₂AlC

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Abstract The compressive properties of ternary compound Cr₂AlC at different temperatures and strain rates were studied. When tested at a strain rate of $5.6 \times 10^{-4} \text{ s}^{-1}$, the compressive strength decreases continuously from $997 \pm 29 \text{ MPa}$ at room temperature to $523 \pm 7 \text{ MPa}$ at 900 °C. The ductile-to-brittle transition temperature is measured to be in the range of 700 to 800 °C. When tested in the strain rate range of 5.6×10^{-5} to $5.6 \times 10^{-3} \text{ s}^{-1}$, Cr₂AlC fails in a brittle mode at room temperature, whereas the deformation mode changes from a brittle to a ductile as the strain rate is lower than $5.6 \times 10^{-4} \text{ s}^{-1}$ when compressed at 800 °C. The compressive strength increases slightly with increasing strain rate at room temperature and it is less dependent on strain rate when tested at 800 °C. The plastic deformation mechanism of Cr₂AlC was discussed in terms of dislocation-related activities, such as kink band formation, delamination, decohesion of grain boundary, and microcrack formation.

Introduction

In recent years, a series of ternary carbides and nitrides, abbreviated as M_{n+1}AX_n (where $n = 1-3$, M is an early transition metal, A is an A group element, and X is C and/or N), have been studied extensively because of their good combination of various properties, such as high mechanical

strength, good electrical and thermal conductivity as well as good chemical stability under oxidation and/or corrosion environments [1–4].

As a member of the M₂AX family, Cr₂AlC was reported by Jeitschko et al. in the 1960s [5]. Nearly 20 years later, Schuster et al. [6] synthesized the compound and identified its lattice parameters and phase relationship in the Cr–Al–C system. Recently, Schneider et al. [7] prepared Cr₂AlC film and studied its elastic properties [8]. Walter et al. [9, 10] fabricated and developed the application of Cr₂AlC film as protective coatings on steel. Lin et al. [11] reported the fabrication, transmission electron microscope characterization [12], and the oxidation behavior of bulk Cr₂AlC [13]. On the other hand, elastic properties of M₂AlC were studied by first principles calculation in recent years [14–18], in which Cr₂AlC was reported to possess high elastic stiffness, though the experimental modulus [8, 13, 19] is significantly lower.

Tian et al. [20] studied the isothermal oxidation behavior of Cr₂AlC ceramics in air at 1100 and 1250 °C for 20 h and found that Cr₂AlC shows good oxidation resistance at the testing temperatures. The electrical and thermal properties of Cr₂AlC at room and elevated temperatures were also reported [19]. Furthermore, its mechanical properties at room temperature such as hardness, elastic modulus, and damage tolerance as well as thermal shock properties were also investigated [21]. However, there is no further report on the mechanical properties, such as high temperature strength and deformation behavior, of Cr₂AlC as far as we know. As a potential candidate of high-temperature materials, the investigation of the mechanical properties at elevated temperature of this compound is a must before its practical application. In this article, therefore, we report the compressive behavior of Cr₂AlC at different temperatures and/or different strain rates.

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Experimental procedure

The starting materials used in this study were chromium (ca. 10 μm , >98%), aluminum (ca. 10 μm , 99.9%), and graphite (ca. 5 μm , 99.7%) (all the materials purchased from Kojundo Chemical Lab., Japan). The powders were weighed according to the designed composition (molar ratio of Cr:Al:C = 2:1.1:1) that was usually used to synthesize Cr_2AlC [19, 20] and mixed by a Turbula shaker mixer for 24 h. The powder mixture was put into a graphite mold and sintered in vacuum using a pulse discharge sintering (PDS) equipment (PAS-V, Sodick Co. Ltd., Japan) at 1250 $^\circ\text{C}$ for 30 min with a heating rate of 50 $^\circ\text{C}/\text{min}$ and a pressure of 50 MPa.

The compressive specimens were cut by an electric discharge machine from the sintered bulk Cr_2AlC . Then, all the specimens were mechanically ground and polished to a dimension of $2.5 \times 2.5 \times 6 \text{ mm}^3$ with smooth surfaces for microscopic observation. The compressive tests were performed with an Instron 8562 universal testing machine at a strain rate ($5.6 \times 10^{-4} \text{ s}^{-1}$) between room temperature and 900 $^\circ\text{C}$, and at different strain rates (5.6×10^{-5} to $5.6 \times 10^{-3} \text{ s}^{-1}$) at room temperature and an elevated temperature of 800 $^\circ\text{C}$, respectively. Three specimens were used for testing at each testing temperature or strain rate. Data were collected as engineering strain, and the recorded load–displacement data during testing were corrected by eliminating the system compliances which were experimentally determined at each individual testing condition (strain rate, temperature). The corrected load–displacement relationship of Cr_2AlC specimens was fairly close to its real value by examining the stress–strain relationship at the elastic part of deformation. For instance, the Young's modulus of the Cr_2AlC specimens measured from such determined stress–strain curve at room temperature at a strain rate of $5.6 \times 10^{-4} \text{ s}^{-1}$ is about 230 GPa, which is reasonably close to the experimental value of 278 GPa [19].

The density of sintered Cr_2AlC sample was measured by Archimedes principle. Phase content was determined by X-ray diffractometry (XRD) (X'pert, Philips, Netherlands) with Cu K α radiation at 30 kV and 40 mA at a scanning speed of 0.02 $^\circ/\text{s}$. The sample was mechanically polished and etched in HF acid for 2 min before scanning electron microscope (SEM) observation. The observations on etched surfaces and fracture surfaces were performed under an SEM (FEI XL30S FEG, Netherlands) equipped with an energy-dispersive spectroscopy (EDS) system.

Results and discussion

Characterization

The XRD pattern of the sintered Cr_2AlC is shown in Fig. 1a, in which all the reflections can be indexed as

Cr_2AlC according to JCPDS9-0017 and the peaks at 51.22, 63.06, and 67.01 degrees as the reflections of Cr_2AlC (105), (107), and (112) respectively as reported in ref. [11]. Measured density of the sample is $5.14 \text{ g}/\text{cm}^3$, which is 98% of the theoretical density. A typical back-scattered SEM image of the etched surface of sintered Cr_2AlC sample is shown in Fig. 1b, in which the gray matrix phase corresponds to Cr_2AlC with a small amount of dark pores inside grains and some bright phases correspond to Cr_7C_3 that were usually found in the synthesized Cr_2AlC . Some dark gray grains that mainly distributed at tri-grain boundaries are identified to be Al_2O_3 according to EDS results (data not shown). This might be a result of the reaction between Al and the absorbed oxygen in starting powder. In addition, Cr_2AlC grains are mostly equiaxed and the average grain size is $6.4 \pm 2.1 \mu\text{m}$.

Compression at different temperatures and strain rates

The relationship between compressive stress and engineering strain in the temperature range of room temperature

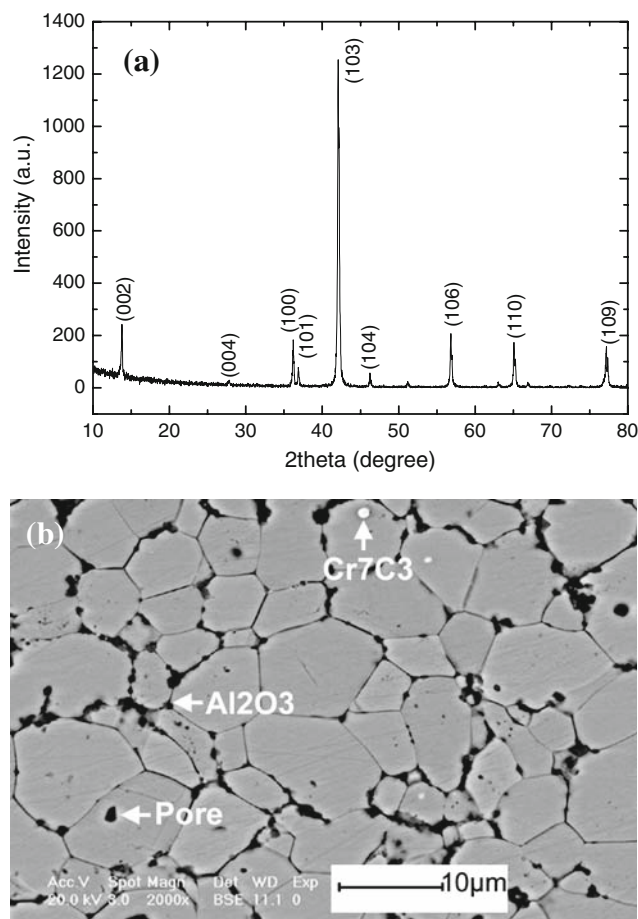


Fig. 1 **a** XRD pattern of Cr_2AlC sample sintered at 1250 $^\circ\text{C}$ for 30 min by PDS. **b** A typical backscattered SEM micrograph image on the surface of polished and etched sample

to 900 °C at a strain rate of $5.6 \times 10^{-4} \text{ s}^{-1}$ is shown in Fig. 2a (for clarity, curves are shifted). It is evident that the Cr_2AlC specimens fail abruptly in a brittle mode after the maximum compressive stress is reached when tested below 700 °C. However, an obviously plastic deformation region is observed on the stress–strain curves when tested at temperatures higher than 800 °C, and this region becomes broadened at 900 °C. In fact, the specimen tested at 900 °C does not break but is deformed to a spindle-like shape. Therefore, the ductile-to-brittle transition temperature of Cr_2AlC locates in the range of 700 to 800 °C. Accordingly, we select two temperatures of room temperature and 800 °C to study the effect of strain rate on the compressive behavior of Cr_2AlC .

The dependence of compressive strength on testing temperature is shown in Fig. 2b. It can be found that the compressive strength of Cr_2AlC specimen decreases continuously from $997 \pm 29 \text{ MPa}$ at room temperature to $749 \pm 21 \text{ MPa}$ at 800 °C and then drops significantly to

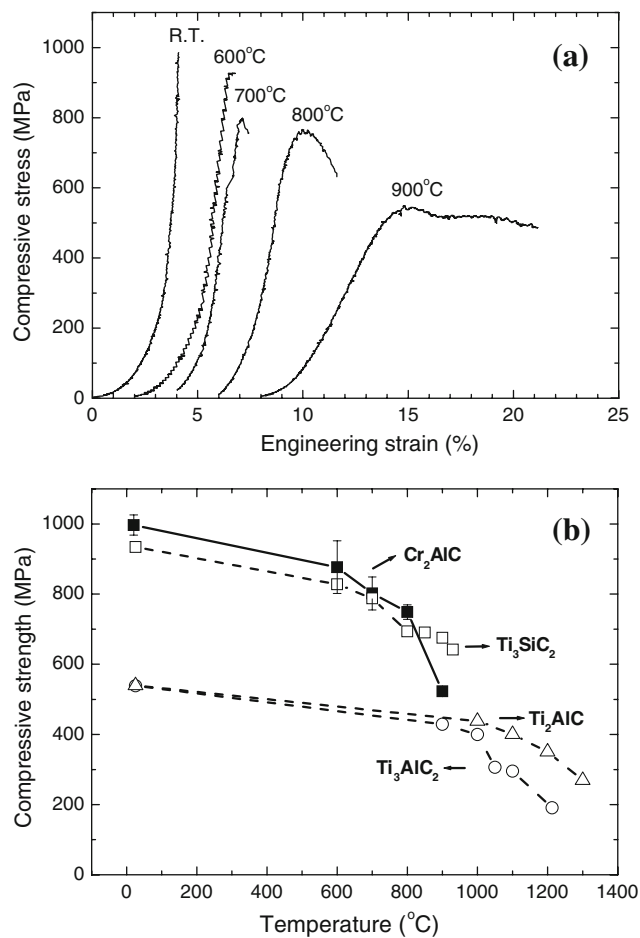


Fig. 2 **a** The relationship between compressive stress and engineering strain. **b** The dependence of compressive strength of Cr_2AlC specimens on the testing temperature. For comparison, the compressive strengths of Ti_3SiC_2 , Ti_3AlC_2 , and Ti_2AlC at different temperatures are also illustrated

$523 \pm 7 \text{ MPa}$ at 900 °C, indicating that the compressive strength in the ductile mode is much more sensitive to temperature than that in the brittle regime. For comparison, the compressive strengths of Ti_3SiC_2 [22], Ti_3AlC_2 [23], and Ti_2AlC [24] at different temperatures are also plotted in Fig. 2b. It can be found that the compressive strengths of Cr_2AlC are not only higher than those of Ti_3AlC_2 and Ti_2AlC , but also slightly higher than those of Ti_3SiC_2 at temperatures below 800 °C. However, it becomes lower than that of Ti_3SiC_2 at the testing temperature of 900 °C, although it is still higher than that of Ti_3AlC_2 and Ti_2AlC . The high compressive strength of Cr_2AlC can be associated with the refined microstructure.

Figure 3a shows the stress–strain curves of Cr_2AlC specimens tested at room temperature at different strain rates. Despite the scatter in maximum compressive stresses recorded at different strain rates, these stress–strain curves of specimens tested at either higher strain rates or lower strain rates are similar to that of specimen tested at a strain rate of $5.6 \times 10^{-4} \text{ s}^{-1}$, which is used as a standard strain rate for testing at different temperatures. This result indicates that the brittle deformation mode does not change in the strain rate range of 5.6×10^{-5} to $5.6 \times 10^{-3} \text{ s}^{-1}$ when tested at room temperature.

The stress–strain curves of Cr_2AlC specimens tested at 800 °C at different strain rates are shown in Fig. 3b. It is evident that the stress–strain curve varies from an elastic way to a plastic one as the strain rate decreases, and the plastic deformation region becomes more broadened as strain rate is below $5.6 \times 10^{-4} \text{ s}^{-1}$, illustrating that the deformation mode changes from a ductile mode to a brittle mode when the strain rate is above $5.6 \times 10^{-4} \text{ s}^{-1}$. Note that the specimens tested at 800 °C at low strain rates, such as 1.4×10^{-4} and $5.6 \times 10^{-5} \text{ s}^{-1}$, do not break but display a spindle-like shape, similar to the specimen tested at 900 °C at a strain rate of $5.6 \times 10^{-4} \text{ s}^{-1}$, revealing that high temperature and low strain rate have similar effect on the plastic deformation of Cr_2AlC .

The dependences of compressive strength of Cr_2AlC specimens on the strain rate at room temperature and at 800 °C are shown in Fig. 3c, in which it can be found that the compressive strength increases gradually with the increase in strain rate at room temperature, while it is less dependent on strain rate when tested at 800 °C.

Microstructural examination on the deformed and fracture surface

At certain strain rate ($5.6 \times 10^{-4} \text{ s}^{-1}$), the deformed and the fracture surfaces of the select but typical samples tested at room temperature and 800 °C are shown in Fig. 4a–d. When tested at room temperature, one main crack propagates through the specimen and makes about a 45° angle

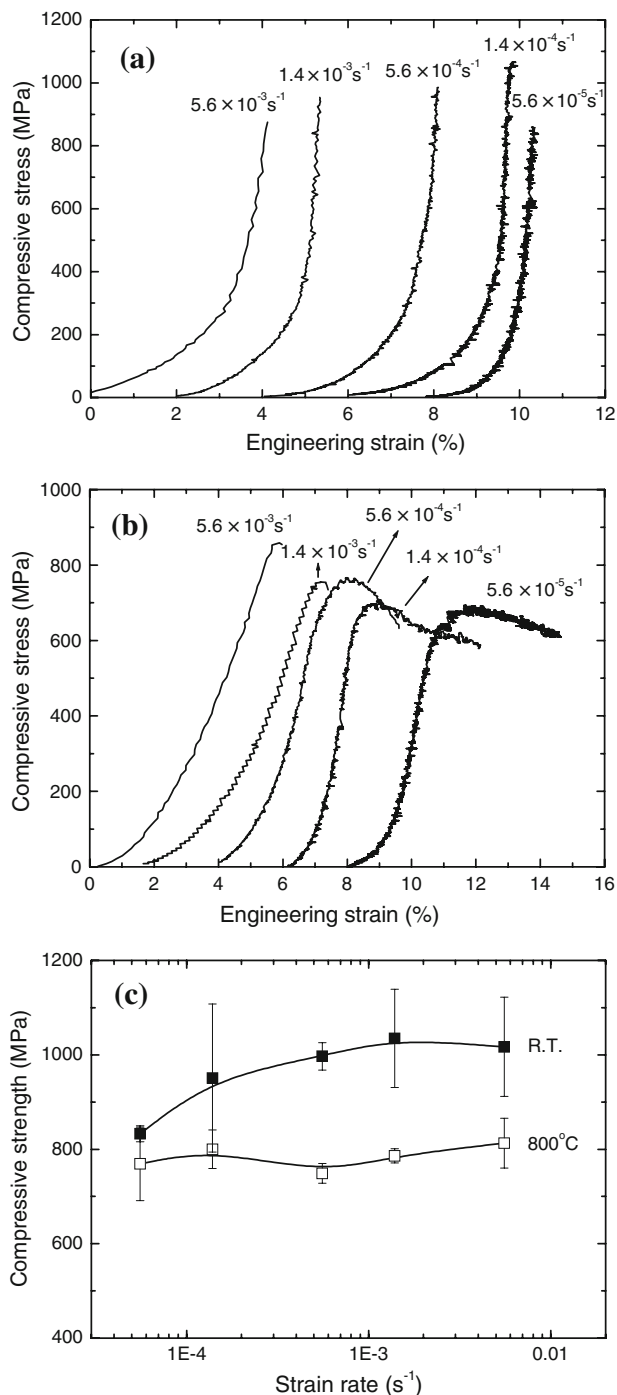


Fig. 3 The stress–strain curves of Cr₂AlC specimens tested at a different strain rate **a** at room temperature and **b** at 800 °C. **c** The dependences of strength on strain rate of Cr₂AlC specimens at room temperature and 800 °C

with the load direction (indicated as white arrow) (Fig. 4a). In contrast, when tested at 800 °C, a large number of cracks appear and most of them are aligned on the load direction (Fig. 4b). In addition, the fracture surfaces of the samples tested at all the testing temperatures are characterized predominantly by intergranular cracking, as

represented by Fig. 4c (room temperature) and Fig. 4d (800 °C). Grain boundary decohesion is observed after compression at high temperature (Fig. 4e). Furthermore, delamination within grains on the fracture surface (Fig. 4f) can be found frequently in the specimens tested at 800 °C.

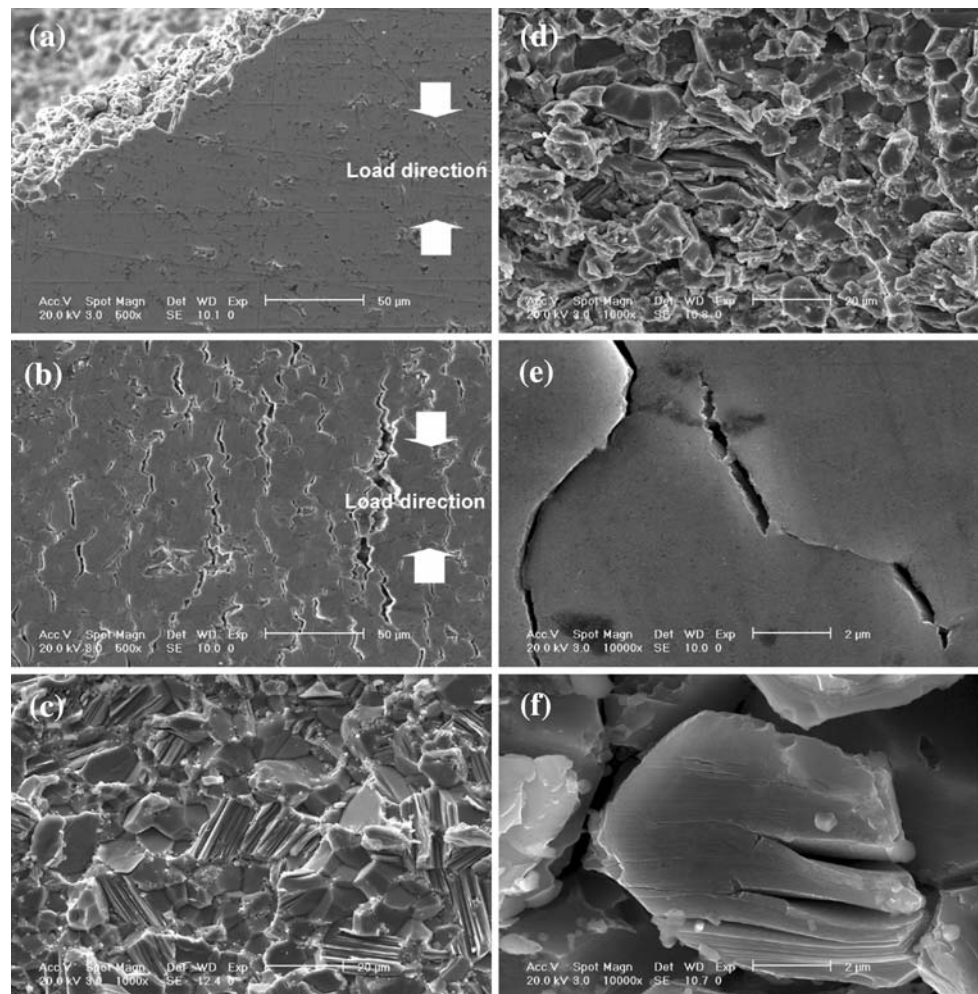
The microstructural images of specimens tested at room temperature at other strain rates are similar to those tested at $5.6 \times 10^{-4} \text{ s}^{-1}$ (Fig. 4a, c), and hence they are not shown. Figure 5 shows the deformed surface and the fracture surface of the sample tested at 800 °C at a high strain rate of $5.6 \times 10^{-3} \text{ s}^{-1}$, in which the deformed region is confined in the vicinity of cracking area (Fig. 5a) and intergranular cracking characteristic, similar to the specimen tested at the same temperature at a strain rate of $5.6 \times 10^{-4} \text{ s}^{-1}$ (Fig. 4d), is also shown (Fig. 5b). Moreover, the kink band within grain is usually observed on the fracture surface, as shown in Fig. 5c.

Effect of temperature and strain rate on compressive properties

The difficult motion of dislocations at low temperatures and the lack of five independent slip systems are responsible for the absence of macro-plasticity in Cr₂AlC at a temperature below 700 °C, i.e., fracture in a brittle mode. By increasing the testing temperature, the motion of dislocations that is believed to be operative only at the basal plane [22–24] becomes readier and they will pile-up against the grain boundaries. The pile-ups would produce microcracks in various ways [25], such as grain buckling, kink band formation, etc. On the other hand, the cohesion strength of grain boundary decreases and the propagation of microcracks along grain boundaries becomes effective with increasing temperature. Furthermore, the microcracks will propagate with increasing applied stress during loading, leading to the grain boundary decohesion (Fig. 4e). It should be pointed out that the small amount of Al₂O₃ phases mostly distributed at grain boundary could also affect the mechanical strengths in two aspects: introducing tensile thermal stress in the Cr₂AlC matrix and weakening the grain boundary.

Note that a creep deformation, dislocation activity or diffusion-controlled creep, should be considered when the compressive tests were carried out at high temperatures. It is known that the diffusion-controlled creep needs to maintain the cohesion between deformed grains for the transport of material either along the grain boundary or through the lattices of grains. However, based on the microstructural examination, it can be found that there are no obvious plastically deformed grains appeared but mostly consisting of intergranular cracking (Fig. 4b, e) and delamination (Fig. 4f) within grains. Therefore, it is reasonable to believe that the dominant high-temperature deformation mechanism in Cr₂AlC is related to the dislocation activities.

Fig. 4 The deformed surface of Cr_2AlC specimens tested **a** at room temperature and **b** 800 °C. The fracture surface of the sample tested **c** at room temperature and **d** 800 °C. **e** Typical grain-boundary decohesion and **f** delamination within grain in the Cr_2AlC specimens tested at 800 °C. All tests at a strain rate of $5.6 \times 10^{-4} \text{ s}^{-1}$



At room temperature, the deformation mode is not sensitive to strain rate because of the difficult motion of dislocation. On the other hand, when tested at 800 °C, the deformation behavior depends on strain rate. This is contributed from more dislocations in the system activated at this elevated temperature. The brittle failure at a strain rate higher than $1.4 \times 10^{-3} \text{ s}^{-1}$ is attributed to insufficient time for the motion of activated dislocations to form microcracks and hence macro-structural plastic deformation.

Generally, the motion of dislocation is not only a thermally activated process, but also a stress-assisted process. Johnston and Gilman [26] reported that strain rate ($\dot{\epsilon}$) is related to the dislocation velocity (v), which is in turn related to the shear stress for dislocation motion (τ) [27] according to the following equations:

$$\dot{\epsilon} = \mathbf{b}n v, \quad (1)$$

$$v = \left(\frac{\tau}{\tau_0} \right)^P \quad (2)$$

where \mathbf{b} is the Burgers vector, n the number of dislocation per unit area, τ_0 and P are experimentally determined

material constants. Accordingly, the shear stress required for dislocation motion increases with the increasing strain rate. Therefore the compressive strength of Cr_2AlC increases with the increase in strain rate at room temperature, although the tendency is not so obvious at 800 °C (Fig. 3c). The mechanism for the less strain rate sensitivity at 800 °C than at room temperature needs further exploration, but it is likely related to the microstructural deformations, such as microcracks, delamination, and kink band.

Conclusion

The compressive strength of Cr_2AlC decreases continuously from $997 \pm 29 \text{ MPa}$ at room temperature down to $523 \pm 7 \text{ MPa}$ at 900 °C and this strength decreasing tendency is more obvious as the testing temperature is higher than 800 °C. Cr_2AlC compound fails in a brittle mode up to 700 °C and a ductile mode at and above 800 °C. At room temperature, the brittle deformation mode does not change

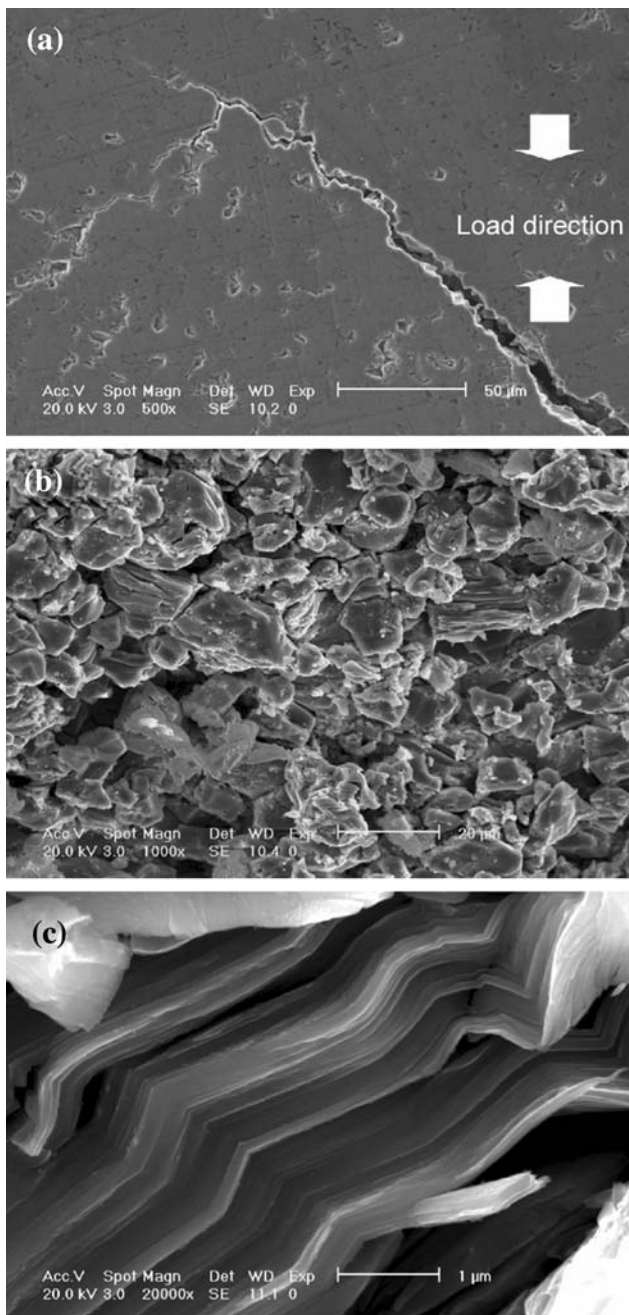


Fig. 5 **a** The deformed surface morphology and **b** the fracture surface of the Cr_2AlC specimen tested at 800°C at a strain rate of $5.6 \times 10^{-3} \text{ s}^{-1}$. **c** Typical kink band within grain on the fracture surface

in the tested strain rate range. At 800°C , however, the deformation mode changes from a ductile mode to a brittle mode when the strain rate is above $5.6 \times 10^{-4} \text{ s}^{-1}$. The

compressive strength is almost independent of the strain rate especially when tested at 800°C . The plastic deformation mechanism of Cr_2AlC can be explained by the dislocation-related activities, such as microcracks and decohesion of grain boundary and delamination and kink band within grains.

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